

## P450cam Visits an Open Conformation in the Absence of Substrate<sup>†,‡</sup>

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**ABSTRACT:** P450cam from *Pseudomonas putida* is the best characterized member of the vast family of cytochrome P450s, and it has long been believed to have a more rigid and closed active site relative to other P450s. Here we report X-ray structures of P450cam crystallized in the absence of substrate and at high and low [K<sup>+</sup>]. The camphor-free structures are observed in a distinct open conformation characterized by a water-filled channel created by the retraction of the F and G helices, disorder of the B' helix, and loss of the K<sup>+</sup> binding site. Crystallization in the presence of K<sup>+</sup> alone does not alter the open conformation, while crystallization with camphor alone is sufficient for closure of the channel. Soaking crystals of the open conformation in excess camphor does not promote camphor binding or closure, suggesting resistance to conformational change by the crystal lattice. This open conformation is remarkably similar to that seen upon binding large tethered substrates, showing that it is not the result of a perturbation by the ligand. Redissolved crystals of the open conformation are observed as a mixture of P420 and P450 forms, which is converted to the P450 form upon addition of camphor and K<sup>+</sup>. These data reveal that P450cam can dynamically visit an open conformation that allows access to the deeply buried active site without being induced by substrate or ligand.

Cytochrome P450s are ubiquitous heme monooxygenases that activate O<sub>2</sub> for oxygen atom insertion into a wide variety of substrates (1). Over 10000 forms of P450s have been identified in bacteria, archaea, plants, fungi, and all higher eukaryotes (2). Examples include major drug metabolizing enzymes of the liver and biosynthetic enzymes involved in steroid and prostaglandin pathways. These enzymes all share a common protein fold first observed for the camphor metabolizing P450cam from *Pseudomonas putida* (3, 4).

The large diversity in the specificity or promiscuity of P450s is believed to be due largely to variations in the structure of the substrate binding channel, which is defined by elements including the F and G helices, FG loop, and regions near the B' helix, which fold over the heme and the I helix to enclose the substrate and position it for attack by the ferryl heme center. Sequence variation of these elements results in P450s with structurally distinct substrate binding channels, and large differences in these regions have been observed in the structures of bacterial (5), microsomal (6), and mitochondrial P450s (7), supporting the view that the different conformations of the F, G, and B' helices are responsible for the large diversity in size, shape, and specificity of the substrate binding cavity.

Conformational dynamics and plasticity are also recognized to play an important role in P450 substrate access and recognition. Both P450 EryK (8) and PikC (9) have been shown to coexist in open and closed states in the absence of substrate. In addition, conformational changes upon substrate binding have been seen for several P450s. Changes were seen in the F and G helices of P450 BM-3 (10) and CYP158A2 (11) or in the BC loop of

CYP119 (12), CYP105P1 (13), and CYP130 (14). More dramatic changes involving both the F and G helices and the BC loop were observed in CYP2B4 (15) upon substrate binding. In other examples, a combination of disorder in the BC loop and retraction of the F and G helices was seen (9, 16). Finally, both the BC loop and C helix were observed to become partially disordered in CYP51 upon ligand association (17).

P450cam is the most well characterized of all P450s, and in many ways it has served as the archetypal model for all P450s, particularly those with high substrate specificity. However, P450cam has remained somewhat of an outlier with regard to the role of conformational flexibility and dynamics, as there has been a general belief that the active site of P450cam is more static and shielded from solvent than other P450s (5). This is largely due to the report of a camphor-free structure obtained by soaking dithiothreitol (DTT)<sup>1</sup> out the active site of crystals, showing a closed conformation that is very similar to the camphor-bound form (3, 4, 18). Small angle X-ray scattering (19) and hydrostatic pressure experiments (20–22) have also been used to support the view that substrate-free P450cam exists in a closed conformation. However, it is clear that the structure must undergo significant conformational changes before substrate can enter the solvent-inaccessible binding site. Thus, either the substrate must induce a conformational change in the protein to initiate its binding or the protein must dynamically visit open conformations in the absence of substrate to allow its entry into the deeply buried channel.

Previous studies have shown that P450cam can be trapped in a range of open conformations in response to binding large tethered adamantane probes (23–25). These substrate analogues bind with the adamantane at the camphor binding site and accommodate the flexible linkers by adopting states in which the

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<sup>‡</sup>The atomic coordinates and structure factors were deposited in the Protein Data Bank under the codes 3L61 (substrate-free, 200 mM [K<sup>+</sup>]), 3L62 (substrate-free, low [K<sup>+</sup>]), and 3L63 (camphor-bound, low [K<sup>+</sup>]).

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<sup>1</sup>Abbreviations: DTT, dithiothreitol; Pdx, putidaredoxin; RMSD, root mean square deviation.

Table 1: Data Collection, Refinement Statistics, and Validation

	data set		
	camphor-free, low [K <sup>+</sup> ]	camphor-free, 200 mM [K <sup>+</sup> ]	camphor-bound, low [K <sup>+</sup> ]
data collection			
wavelength (Å)	0.979	0.979	0.979
unit cell <i>a</i> , <i>b</i> , <i>c</i> (Å)	65.6, 73.5, 91.9	65.6, 73.8, 92.4	35.4, 98.2, 54.2; $\beta = 103.9^\circ$
space group	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub>
resolution range (Å)	49.0–1.70 (1.79–1.70) <sup>a</sup>	46.2–1.50 (1.58–1.50) <sup>a</sup>	27.8–1.50 (1.58–1.50) <sup>a</sup>
no. of total reflections	175296	506840	211818
no. of unique reflections	49496	72274	57384
completeness (%)	99.8 (99.9)	99.7 (100.0)	100.0 (100.0)
<i>R</i> <sub>merge</sub> (%)	6.2 (41.7)	6.0 (54.1)	6.6 (52.9)
<i>I</i> / $\sigma$ ( <i>I</i> )	13.6 (2.9)	19.4 (3.9)	12.8 (2.3)
Wilson <i>B</i> -value (Å <sup>2</sup> )	17.8	17.6	14.2
refinement statistics			
resolution range (Å)	10–1.7	10–1.5	10–1.5
no. of reflections used	46706	68435	54283
free <i>R</i> reflections (%)	5.0	5.0	5.0
<i>R</i> / <i>R</i> <sub>free</sub>	0.190/0.224	0.197/0.216	0.164/0.196
rmsd bond length (Å)	0.0121	0.0093	0.0096
rmsd bond angle (deg)	1.377	1.284	1.407
Ramachandran analysis (%)			
no. of residues in favored regions	98.5	99.0	98.5
no. of residues in allowed regions	1.5	1.0	1.5
no. of residues in outlier regions	0.0	0.0	0.0
PDB entry	3L62	3L61	3L63

<sup>a</sup>Data for the outermost shell are given in parentheses.

F and G helices are retracted from their position above the closed substrate binding cavity. It was suggested that these open structures represent substrate access pathways into the buried active site, but it has not been clear if these open conformations are relevant to native function or if they represent unnatural states held open by the presence of the artificial ligand.

Here, we report the crystal structures of P450cam crystallized in the absence of camphor and at high and low [K<sup>+</sup>] and in the presence of camphor at low [K<sup>+</sup>]. Both of the camphor-free structures are observed in an open conformation that is distinctly different from the previously observed closed camphor-free state. The implications of this open state to conformational dynamics associated with substrate recognition and binding are discussed.

## EXPERIMENTAL PROCEDURES

**Protein Preparation.** The pETCAM-C334A vector was constructed by inserting P450cam (a gift from Prof. T. Poulos) DNA into the pET15b vector (Novagen) between *Nco*I and *Xho*I restriction enzyme sites and introducing the C334A mutant using QuickChange (Stratagene). This construct expresses a full-length protein with a mutation at Cys334 to Ala. C334A mutation has been shown to increase protein stability by blocking intermolecular dimerization but does not affect protein activity (26). P450cam (C334A) was overexpressed in *Escherichia coli* BL21-(DE3) from vector pETCAM-C334A grown for 24 h at 25 °C after induction by 0.4 mM IPTG and addition of 1 mM 5-amino-levulinic acid. Cell pellets were resuspended in 50 mM KPi, pH 6.0, 1 mM camphor, 0.1 mM heme, 0.5 mg/mL lysozyme, and 50 μg/mL DNase before lysis in a French press. Cell lysates, clarified by centrifugation, were purified on a DEAE anion-exchange column, which was initially equilibrated with 50 mM KPi, pH 6.0, and 1 mM camphor. The column was washed with the same buffer plus 50–100 mM KCl, and the protein was eluted at 300 mM KCl. The pooled fractions were concentrated by

ultrafiltration before loading onto a Sephacryl S-200 size exclusion column equilibrated with 50 mM KPi, pH 6.0, and 300 mM KCl. The protein fractions with *A*<sub>417</sub>/*A*<sub>280</sub> above 1.45 were pooled and exchanged into 50 mM KPi, pH 6.0, 1 mM camphor, and 30 mM β-mercaptoethanol and frozen at –80 °C for further experiments.

**Crystallization, Data Collection, and Crystal Structure Determination.** Camphor was removed from protein stock solutions by buffer exchange into 50 mM Tris, pH 7.4, by gel filtration through two sequential PD-10 columns (GE Healthcare). For camphor-bound P450cam, 2 mM camphor was mixed with 1 mM P450cam before crystallization. Crystals of P450cam were grown by sitting-drop vapor diffusion at 6 °C. Substrate-free crystals grew from 50 mM Tris, pH 7.4, and 12–22% polyethylene glycol 8000 with and without 200 mM K<sup>+</sup>. Camphor-bound crystals grew from 50 mM Tris, pH 7.4, 12–22% polyethylene glycol 8000, and 2–4 mM camphor. The crystals were transferred to cryoprotectant buffer consisting of 50 mM Tris, pH 7.4, 18% polyethylene glycol 8000, and 25% polyethylene glycol 600 with and without 200 mM K<sup>+</sup>, mounted on nylon loops, and flash frozen at 77 K.

X-ray diffraction data were collected at 100 K using beamline 7-1 or 11-1 at the Stanford Synchrotron Radiation Laboratory. Data were processed using Scala (27). Initial models for the closed and open structures were derived from those of camphor-bound P450cam (PDB entry 5CP4) and the most open conformation for a series of tethered substrate analogues (unpublished results), respectively. Molecular replacement was done using Molrep (28), and model fitting and refinement were done with Coot (29) and Refmac5 (30). The final models were validated using the programs Procheck (31), Scheck (32), and Molprobit (33). Statistics for data collection and refinement are shown in Table 1. The atomic coordinates and structure factors were deposited in the Protein Data Bank under the codes 3L61, 3L62,

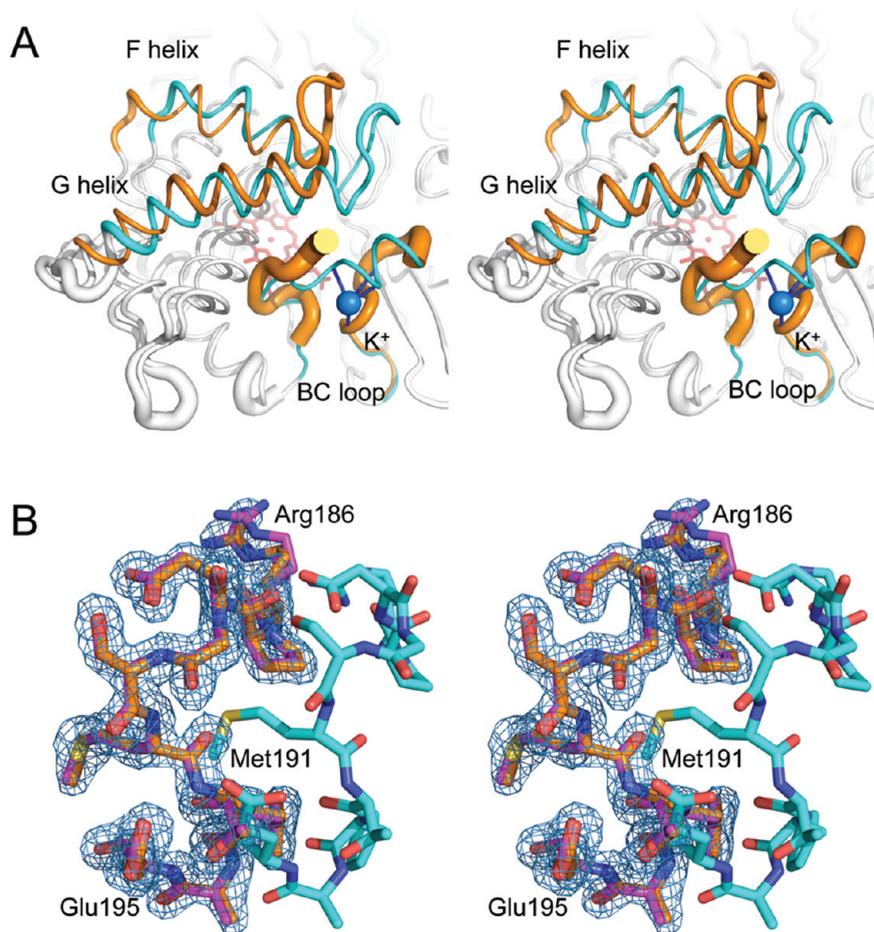


FIGURE 1: Superimposed structures of the closed (PDB entry 3L63) and the open (PDB entry 3L62) forms of P450cam. (A) Stereoview of the two conformational forms showing  $B$ -values as backbone tube thickness. Only the F and G helices and BC loop are colored. The closed and open forms are shown in cyan and orange, respectively. A blue sphere shows  $K^+$ . (B)  $2F_o - F_c$  electron density map contoured at  $1\sigma$  of the FG loop for the open structure (PDB entry 3L62). The corresponding model is shown in orange, overlaid on the models for the other open conformation (PDB entry 3L61) in magenta and the closed conformation (PDB entry 3L63) in cyan.

and 3L63. All structural graphics were generated using the Pymol Molecular Graphics System (34).

**UV-Visible Spectroscopy.** Open form crystals, which were grown in the absence of camphor, were initially dissolved in 50 mM Tris, pH 7.4. The redissolved protein samples were divided into two aliquots. One aliquot was diluted into 50 mM Tris, pH 7.4, while the other was diluted into 50 mM Tris, pH 7.4, 0.5 mM camphor, and 200 mM KCl. UV-visible spectra of ferric and ferrous CO-bound forms were recorded on a HP 8453 UV-visible spectrophotometer. For spectra of the ferrous CO bound form, the protein sample was bubbled with CO gas before addition of aliquots of a fresh solution of sodium dithionite (5–10 mM final). For single crystal spectroscopy, absorbance spectra were recorded of crystals mounted on nylon loops in an Oxford cryostream operating at 100 K using an in-house system based on an Ocean Optics USB4000 spectrometer.

## RESULTS

P450cam crystallized in the absence of camphor or other distal heme ligands and at low concentration of monovalent cations was observed in an open conformation characterized by retraction of F and G helices and FG loop and loss of order in the B' helix and bound  $K^+$  (Figure 1A). The rmsd between the closed and open conformation was 1.8 Å with the greatest displacement at the FG loop (Figure 2), which showed a 9 Å retraction relative

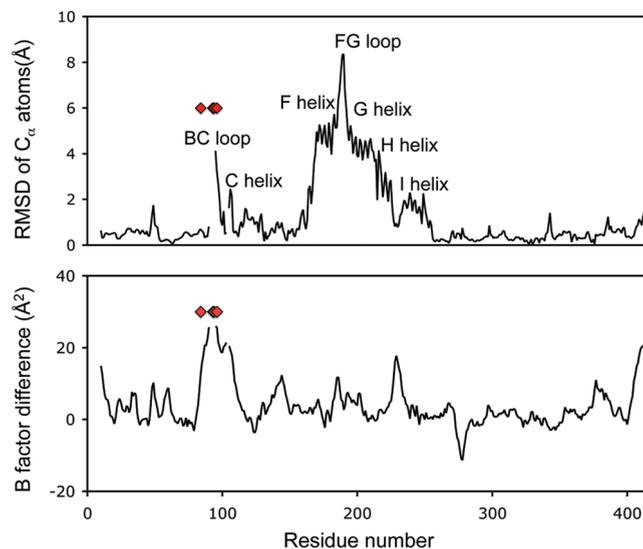


FIGURE 2: Per residue rmsd and  $B$  factor difference between the open and closed conformations of P450cam.  $K^+$  coordination residues are marked by red diamonds.

to the closed camphor-bound state, while the F and G helices were displaced by about 5–6 Å. The FG loop showed well-defined electron density (Figure 1B), and the large movements of

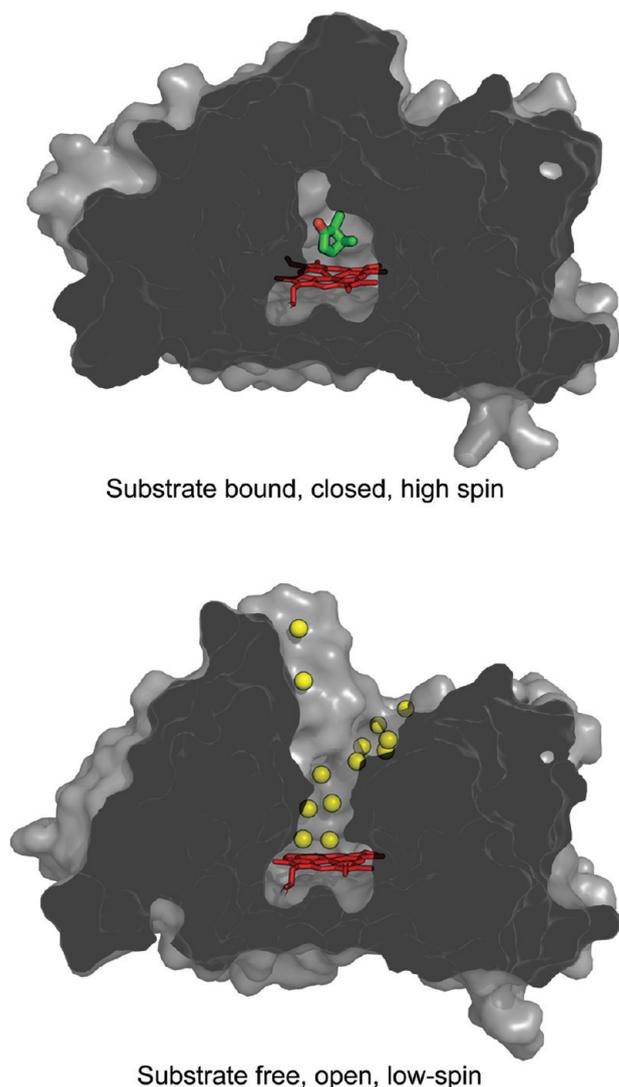


FIGURE 3: Cross-section view of the solvent-accessible surface calculated for the closed (PDB entry 3L63) and the open (PDB entry 3L62) forms of P450cam. Heme and camphor at the active site are shown in red and green, respectively. Ordered waters seen in the channel are shown as yellow spheres.

this segment were not associated with significantly increased  $B$ -values (Figure 2). In contrast, the B' helix, comprising residues 91–94, and the  $K^+$  ion were completely disordered in the open camphor-free state.

Additional structures were solved to determine the individual roles played by camphor and  $K^+$  ion in the conformational state of the enzyme. The 1.5 Å structure obtained from a crystal grown in the absence of camphor but in the presence of 200 mM  $K^+$  was essentially identical to the open state seen at low  $[K^+]$  and did not result in increased ordering of the B' helix, occupation of the  $K^+$  site, or closure of the active site channel (Figure 1B). A third structure at 1.5 Å of P450cam crystallized in the presence of 2 mM camphor and at low  $[K^+]$  was observed in a completely closed conformation (Figures 1 and 3), with well-defined density for camphor at the active site. In addition, electron density is observed at the  $K^+$  binding site of the B' helix in spite of attempts to exclude monovalent cations during crystallization. When the structural model was refined with a full occupancy of  $K^+$  at the coordination site, the  $F_o - F_c$  map showed negative electron density around the  $K^+$ . Thus, the site might be occupied by  $Na^+$  or  $K^+$  at lower occupancy. Nonetheless, this structure was essen-

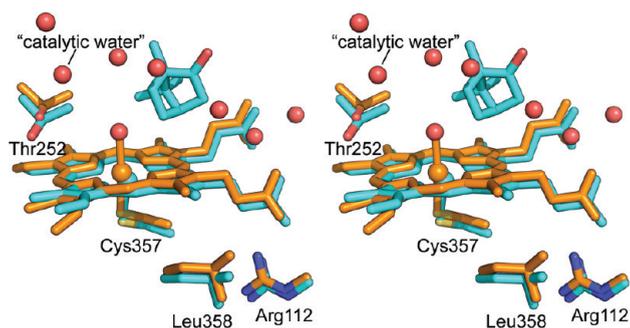


FIGURE 4: Movements of the heme and nearby residues in the open conformation (PDB entry 3L62) relative to the closed form (PDB entry 3L63). The closed and open forms are colored in cyan and orange, respectively. Camphor is only present in the closed form. Waters shown in red are only present in the open form. In the closed camphor-bound state the heme is 5-coordinate with only Cys357 as proximal ligand. In the open conformations the heme is 6-coordinate, with a distally coordinated water shown bound to the iron.

tially identical to the previously reported structure of camphor-bound enzyme (3, 4). We also considered the possibility that soaking crystals of the open conformation in high concentrations (8 mM) of camphor might reveal camphor bound to the open conformation. However, no significant difference density was seen in the soaked crystals (data not shown), indicating that camphor has little intrinsic affinity for the open conformation.

The open camphor-free conformation results in a deep water-filled channel, exposing the substrate binding site to solvent (Figure 3). At least 12 ordered water molecules are observed within the open channel which are excluded in the closed camphor-bound state. Included in this set of ordered waters are the distal heme-bound water responsible for substrate-induced spin-state conversion and a water near the position of the "catalytic" water observed in the  $Fe^{2+}-O_2$  complex (35, 36) (Figure 4). Accommodation of this "catalytic" water is possible due to the larger space in the distal cavity and a small rotation of the Thr252 side chain. The locations of these channel waters are distinct from those previously observed to occupy the camphor site in the closed camphor-free form (18). However, an unidentified small peak at  $2\sigma$  was seen at the camphor site that was too large for a water and too small for camphor. Possible candidates for this density are Tris and  $\beta$ -mercaptoethanol from the original protein stock and the crystallization buffer. The smaller  $\beta$ -mercaptoethanol fits better to the electron density, but the identity of this peak remains uncertain. Additional electron density was also observed at  $\sim 1\sigma$  throughout the open channel that may represent disordered solvent.

As shown in Figure 4, subtle structural differences between the open and closed conformations extend to the proximal heme face. Both the Arg112 and Leu358 side chains move slightly toward the heme in the open conformation, making contact with the heme D ring. Small changes were also seen in the geometry of the heme which shows increased ruffling. The iron in the open conformation moves 0.7 Å toward the heme plane, and the axial Cys ligand follows this movement. While the Fe–S bond length was refined to be  $\sim 0.05$  Å shorter in the open relative to the closed conformations (Table 2), this difference may be within the uncertainty of refinement precision. Overall, these proximal changes are very similar in nature to those seen in the L358P mutant which was proposed to mimic changes induced by Pdx binding (37).

Table 2: Geometry around the Heme of P450cam

	structure		
	camphor-bound, low [K <sup>+</sup> ]	camphor-free, low [K <sup>+</sup> ]	camphor-free, 200 mM [K <sup>+</sup> ]
Fe–OH <sub>2</sub> (Å)	NA	2.25	2.17
Fe–S (Å)	2.38 (2.37) <sup>a</sup>	2.31	2.33
Fe–NA (Å)	2.07 (2.05)	2.02	2.04
Fe–NB (Å)	2.04 (2.05)	2.04	2.01
Fe–NC (Å)	2.08 (2.05)	2.04	2.07
Fe–ND (Å)	2.11 (2.06)	2.05	2.05
∠NA–Fe–S (deg)	100.4 (102.3)	96.9	96.8
∠NB–Fe–S (deg)	90.1 (92.3)	88.5	88.8
∠NC–Fe–S (deg)	94.8 (93.9)	88.1	87.4
∠ND–Fe–S (deg)	105.6 (103.5)	96.1	95.6

<sup>a</sup>Data from structure of PDB entry 2CPP (4) is shown in parentheses.

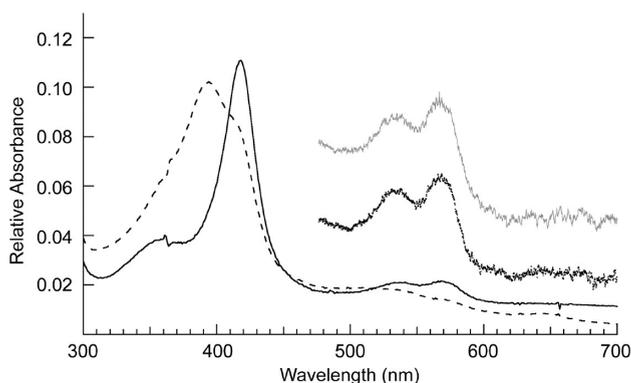


FIGURE 5: UV–visible spectra of P450cam derived from open form crystals. Single crystal UV–visible spectra at 100 K of P450cam grown in camphor-free buffer 50 mM Tris, pH 7.4, and 200 mM KCl (black dots) and grown in camphor-free buffer 50 mM Tris, pH 7.4 (gray dots). Spectra are also shown for redissolved open form crystals in 50 mM Tris, pH 7.4 (solid line), and in 50 mM Tris, pH 7.4, 200 mM KCl, and 0.5 mM camphor (dashed line).

Spectra of frozen single crystals at 100 K of P450cam crystallized in camphor-free, low [K<sup>+</sup>], and in camphor-free, 200 mM [K<sup>+</sup>] conditions both exhibit  $\alpha/\beta$  bands at 568 and 535 nm which are characteristic of the low-spin state expected from the observed water coordination to heme (Figure 5). When these open-state crystals are redissolved in 50 mM Tris, pH 7.4, they remain in a low-spin state, while addition of camphor (0.5 mM final) and K<sup>+</sup> (200 mM final) to this solution causes conversion to high spin, as seen by the significant attenuation of the  $\alpha/\beta$  bands, appearance of the charge transfer band at 645 nm, and a shift of the Soret from 418 to 393 nm (Figure 5). This shows that the open form of P450cam seen in the crystal is capable of binding camphor to reversibly generate the closed camphor-bound state. Finally, the ferrous CO complex produced from open state crystals redissolved in 50 mM Tris, pH 7.4, without camphor or K<sup>+</sup> is observed to exist as a mixture of P450 and P420 forms, while addition of 0.5 mM camphor and 200 mM K<sup>+</sup> to this sample of redissolved crystals shows that it has converted to predominantly the P450 form (Figure 6).

## DISCUSSION

A number of significant conclusions can be made from the simple observation that P450cam crystallized in the absence of camphor exists in an open conformation.

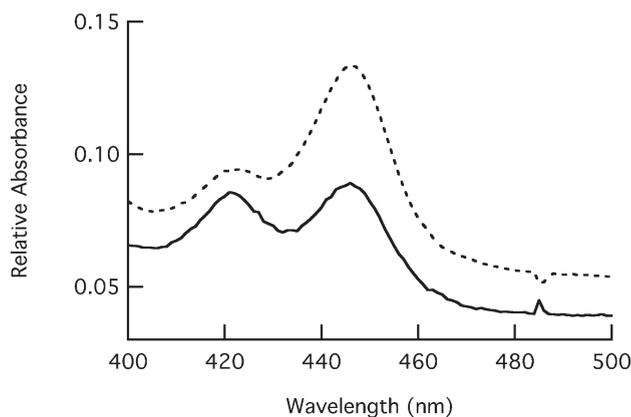


FIGURE 6: UV–visible spectra of the ferrous CO complex for P450cam derived from open form crystals. The open form crystals were dissolved in 50 mM Tris, pH 7.4 (solid line), or in 50 mM Tris, pH 7.4, 200 mM KCl, and 0.5 mM camphor (dashed line) before conversion to the ferrous CO complex.

The open conformation described here would appear to be at variance with the previous report of the substrate-free structure which was observed in a closed conformation (18). However, it is noted that this closed camphor-free structure was obtained by first crystallizing the protein with DTT bound to the heme in place of camphor and subsequently soaking DTT out of the crystals. As such, it is possible that the closed DTT-bound conformation was trapped by the crystal lattice, preventing its conversion to the open conformation. Recently, a closed P450cam structure partially bound to camphor was reported, but this crystal was also grown with excess DTT in addition to 1 mM camphor and 200 mM K<sup>+</sup> (38). Consistent with this, we find in this study that crystals of the open camphor-free protein, when soaked in excess camphor, fail to convert to the closed conformation, which is observed when the same protein is crystallized in the presence of camphor. Again, this suggests that crystal packing resists the conversion between open and closed conformations.

The open camphor-free conformation is very similar to that observed when bound to large tethered substrate analogues containing linkers that prevent simultaneous binding and closure of the channel (23–25). The rmsd between the camphor-free structure reported here and the most open of these tethered substrate-bound forms is 0.5 Å (Figure 7). The similarity of these structures shows that the protein samples the same open conformation in the absence of any substrate as when it is bound to the long tethered substrates, showing that the open conformation is not the result of artificially forcing the channel open. Recent observations suggest that CYP119 exists in multiple conformational states that depend on heme coordinating inhibitors, and these multiple conformations persist even at saturating ligand concentrations. This suggests that both ligand association and dissociation are governed by protein conformational dynamics (39, 40).

P450cam is the only P450 known to have a specific K<sup>+</sup> binding site in the B' helix. K<sup>+</sup> has been shown to enhance camphor binding by 10-fold (41) and to increase the stability of the enzyme (22, 42). However, crystallization in the presence of high [K<sup>+</sup>] alone does not induce any change in the open conformation. This suggests that K<sup>+</sup> binding does not in itself provide sufficient stability to reorder the B' helix, and its presence alone is not sufficient to drive closure of the active site channel in the absence of camphor. Thus the ordering of the B' helix and K<sup>+</sup> coordination

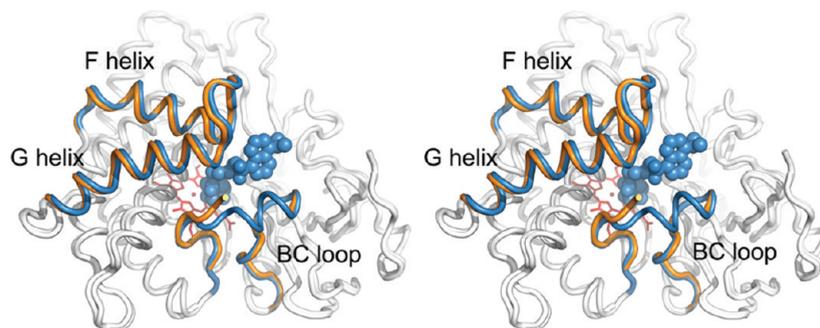


FIGURE 7: Stereoview of the superimposed structures of P450cam in the substrate-free open conformation (orange) (PDB entry 3L62) and the previously reported open state bound to a tethered substrate (blue) (PDB entry 1RF9). Only the F and G helices and BC loop are colored. The tethered substrate, adamantane-1-carboxylic acid 5-dimethylaminonaphthalene-1-sulfonylaminobutylamide, is shown as a CPK model in blue. The most significant difference in the structures is the increased disorder in the B' helix for the substrate-free form.

appear to be driven by substrate binding. As the  $K^+$  binding affinity has been estimated to be very weak even in the presence of camphor ( $\sim 10$  mM) (41), we initially assumed that two successive gel filtration steps would result in its removal from the camphor-bound state. However, the structure of P450cam crystallized in the presence of camphor and at low  $[K^+]$  is observed in a closed conformation with the  $K^+$  site partially occupied, showing that a monovalent cation is still sequestered by the protein or scavenged from the media. This suggests that camphor-driven closure of the substrate channel accompanies cation binding and reordering of the B' helix, and it thus provides a structural explanation for the known synergistic effect of camphor and  $K^+$  binding (41).

In spite of the hydrophobic nature of the exposed substrate binding channel in the open conformation, at least 12 ordered waters and evidence for significantly more disordered solvent molecules are seen in the open conformation of P450cam. This is consistent with the previous report that the volume change induced by osmotic pressure involves reorganization of approximately 19 waters during spin-state conversion (20).

Comparison of the open and closed conformations suggests that multiple small changes propagate to the proximal region near the proposed Pdx binding site that may have functional significance. These changes are very similar to those seen for the L358P mutant, which mimics conformational changes seen upon Pdx binding (37, 43). The L358P has 30 times higher binding affinity for Pdx than the WT P450cam (44) and hydroxylates camphor using sodium dithionite as the sole reductant (45). Pdx binding has been shown to induce changes in the spin state (46) and electronic structure (47) of P450cam and triggers trapping of water molecules in the P450cam–Pdx complex (48, 49). It has also been shown that Pdx binding causes perturbations at the B' helix (50) and cis–trans isomerization of the Ile88–Pro89 peptide bond (51). Although we did not see such a transition of the Ile88–Pro89 peptide bond in the open-state crystal structures, given the similarity of changes near the proximal site between the L358P mutant and WT compared with the changes between the open and closed form, it is possible that Pdx may modulate structural changes that are coupled to the open–closed transition.

When crystals of the open conformation of P450cam are dissolved in the absence of camphor and at low  $[K^+]$  and converted to the ferrous CO complex, a mixture of P420 and P450 species is seen, while addition of camphor and  $K^+$  results in predominantly the P450 form. It is reported that high pressure reversibly converts P450 BM3 and CYP119 to a

P420 form that is associated with a change in molecular volume and spin-state conversion (52, 53). For P450cam, intermediate pressure induces a spin-state change that is reversible, but higher pressure induces an irreversible conversion to P420 (54, 55). However, a structural basis for these changes has not been understood. Our results suggest that the open conformation obtained by crystallization of camphor-free enzyme may be associated with a P420 form that reverts to the closed P450 form upon binding substrate. It has been proposed that changes in iron–cysteine coordination may be responsible for P420 forms (56–58), but it is unclear at present if the slightly shorter Fe–S distance seen in the open conformations is significant in this regard (Table 2).

The observation of an open conformation of P450cam unifies a developing consensus that conformational dynamics plays an important role in substrate recognition for most if not all P450s. Recent structural studies have also shown that substrate-free P450 EryK can exist in both open or closed conformations in different salt conditions (8), and two monomers in the asymmetric unit of substrate-free P450 PikC displayed alternate open and closed conformations (9). In both cases, the open state resulted from a retraction of the F and G helices similar to the changes seen for P450cam.

The structures reported here of P450cam in the presence and absence of camphor and  $K^+$  ion provide a new picture of substrate access and conformational mobility in this archetypal enzyme. The active site of the closed camphor-bound conformation is solvent inaccessible, requiring a protein conformational change to allow substrate binding and product release. However, the previously reported structure of a camphor-free form was seen in a closed conformation (18), and the only open forms observed for this enzyme have resulted from binding of large tethered substrate analogues (23–25). This has given credence to the idea that the substrate channel opens rarely on its own or requires a secondary camphor binding site on the protein surface as the initial step in a substrate-induced mechanism of channel opening (59).

In summary, it now appears that P450cam, along with many other members of the P450 family, displays a dynamic personality and has revealed that the conformational space sampled by the protein includes an open conformation allowing substrate access to the buried active site.

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## REFERENCES

- Ortiz de Montellano, P. R. (2004) *Cytochrome P-450: Structure, Mechanism, and Biochemistry*, 3rd ed., Kluwer Academics/Plenum, New York.
- Nelson, D. R. (2009) The cytochrome p450 homepage. *Hum. Genomics* 4, 59–65.
- Poulos, T. L., Finzel, B. C., Gunsalus, I. C., Wagner, G. C., and Kraut, J. (1985) The 2.6-Å crystal structure of *Pseudomonas putida* cytochrome P-450. *J. Biol. Chem.* 260, 16122–16130.
- Poulos, T. L., Finzel, B. C., and Howard, A. J. (1987) High-resolution crystal structure of cytochrome P450cam. *J. Mol. Biol.* 195, 687–700.
- Pylypenko, O., and Schlichting, I. (2004) Structural aspects of ligand binding to and electron transfer in bacterial and fungal P450s. *Annu. Rev. Biochem.* 73, 991–1018.
- Johnson, E. F., and Stout, C. D. (2005) Structural diversity of human xenobiotic-metabolizing cytochrome P450 monooxygenases. *Biochem. Biophys. Res. Commun.* 338, 331–336.
- Annalora, A. J., Goodin, D. B., Hong, W., Zhang, Q., Johnson, E. F., and Stout, C. D. (2009) The crystal structure of CYP24A1, a mitochondrial cytochrome P450 involved in vitamin D metabolism. *J. Mol. Biol.* 396, 441–451.
- Savino, C., Montemiglio, L. C., Sciarra, G., Miele, A. E., Kendrew, S. G., Jemth, P., Gianni, S., and Vallone, B. (2009) Investigating the structural plasticity of a cytochrome P450: Three dimensional structures of P450 EryK and binding to its physiological substrate. *J. Biol. Chem.* 284, 29170–29179.
- Sherman, D. H., Li, S., Yermalitskaya, L. V., Kim, Y., Smith, J. A., Waterman, M. R., and Podust, L. M. (2006) The structural basis for substrate anchoring, active site selectivity, and product formation by P450 PikC from *Streptomyces venezuelae*. *J. Biol. Chem.* 281, 26289–26297.
- Li, H., and Poulos, T. L. (1997) The structure of the cytochrome p450BM-3 haem domain complexed with the fatty acid substrate, palmitoleic acid. *Nat. Struct. Biol.* 4, 140–146.
- Zhao, B., Guengerich, F. P., Bellamine, A., Lamb, D. C., Izumikawa, M., Lei, L., Podust, L. M., Sundaramoorthy, M., Kalaitzis, J. A., Reddy, L. M., Kelly, S. L., Moore, B. S., Stec, D., Voehler, M., Falck, J. R., Shimada, T., and Waterman, M. R. (2005) Binding of two flavin substrate molecules, oxidative coupling, and crystal structure of *Streptomyces coelicolor* A3(2) cytochrome P450 158A2. *J. Biol. Chem.* 280, 11599–11607.
- Park, S. Y., Yamane, K., Adachi, S., Shiro, Y., Weiss, K. E., Maves, S. A., and Sligar, S. G. (2002) Thermophilic cytochrome P450 (CYP119) from *Sulfolobus solfataricus*: High resolution structure and functional properties. *J. Inorg. Biochem.* 91, 491–501.
- Xu, L. H., Fushinobu, S., Ikeda, H., Wakagi, T., and Shoun, H. (2009) Crystal structures of cytochrome P450 105P1 from *Streptomyces avermitilis*: Conformational flexibility and histidine ligation state. *J. Bacteriol.* 191, 1211–1219.
- Ouellet, H., Podust, L. M., and de Montellano, P. R. O. (2008) *Mycobacterium tuberculosis* CYP130: Crystal structure, biophysical characterization, and interactions with antifungal azole drugs. *J. Biol. Chem.* 283, 5069–5080.
- Scott, E. E., He, Y. A., Wester, M. R., White, M. A., Chin, C. C., Halpert, J. R., Johnson, E. F., and Stout, C. D. (2003) An open conformation of mammalian cytochrome P450 2B4 at 1.6-Å resolution. *Proc. Natl. Acad. Sci. U.S.A.* 100, 13196–13201.
- Zerbe, K., Pylypenko, O., Vitali, F., Zhang, W., Rousset, S., Heck, M., Vrijbloed, J. W., Bischoff, D., Bister, B., Süßmuth, R. D., Pelzer, S., Wohlleben, W., Robinson, J. A., and Schlichting, I. (2002) Crystal structure of OxyB, a cytochrome P450 implicated in an oxidative phenol coupling reaction during vancomycin biosynthesis. *J. Biol. Chem.* 277, 47476–47485.
- Podust, L. M., Yermalitskaya, L. V., Lepesheva, G. I., Podust, V. N., Dalmaso, E. A., and Waterman, M. R. (2004) Estradiol bound and ligand-free structures of sterol 14 $\alpha$ -demethylase. *Structure* 12, 1937–1945.
- Poulos, T. L., Finzel, B. C., and Howard, A. J. (1986) Crystal structure of substrate-free *Pseudomonas putida* cytochrome P-450. *Biochemistry* 25, 5314–5322.
- Lewis, B. A., and Sligar, S. G. (1983) Structural studies of cytochrome P-450 using small angle x-ray scattering. *J. Biol. Chem.* 258, 3599–3601.
- Di Primo, C., Deprez, E., Hoa, G. H., and Douzou, P. (1995) Antagonistic effects of hydrostatic pressure and osmotic pressure on cytochrome P-450cam spin transition. *Biophys. J.* 68, 2056–2061.
- Fisher, M. T., Scarlata, S. F., and Sligar, S. G. (1985) High-pressure investigations of cytochrome P-450 spin and substrate binding equilibria. *Arch. Biochem. Biophys.* 240, 456–463.
- Hui Bin Hoa, G., and Marden, M. C. (1982) The pressure-dependence of the spin equilibrium in camphor-bound ferric cytochrome-P-450. *Eur. J. Biochem.* 124, 311–315.
- Dmochowski, I. J., Crane, B. R., Wilker, J. J., Winkler, J. R., and Gray, H. B. (1999) Optical detection of cytochrome P450 by sensitizer-linked substrates. *Proc. Natl. Acad. Sci. U.S.A.* 96, 12987–12990.
- Dunn, A. R., Dmochowski, I. J., Bilwes, A. M., Gray, H. B., and Crane, B. R. (2001) Probing the open state of cytochrome P450cam with ruthenium-linker substrates. *Proc. Natl. Acad. Sci. U.S.A.* 98, 12420–12425.
- Hays, A. M. A., Dunn, A. R., Chiu, R., Gray, H. B., Stout, C. D., and Goodin, D. B. (2004) Conformational states of cytochrome P450cam revealed by trapping of synthetic molecular wires. *J. Mol. Biol.* 344, 455–469.
- Nickerson, D. P., and Wong, L. L. (1997) The dimerization of *Pseudomonas putida* cytochrome P450cam: Practical consequences and engineering of a monomeric enzyme. *Protein Eng.* 10, 1357–1361.
- Evans, P. R. (1993) Data reduction. *CCP4 Study Weekend Proceedings DL/SCI/R34*, 114–122.
- Vagin, A., and Teplyakov, A. (1997) MOLREP: An automated program for molecular replacement. *J. Appl. Crystallogr.* 30, 1022–1025.
- Emsley, P., and Cowtan, K. (2004) Coot: Model-building tools for molecular graphics. *Acta Crystallogr., Sect. D* 60, 2126–2132.
- Murshudov, G. N., Vagin, A. A., and Dodson, E. J. (1997) Refinement of macromolecular structures by the maximum-likelihood method. *Acta Crystallogr., Sect. D* 53, 240–255.
- Laskowski, R. A., MacArthur, M. W., Moss, D. S., and Thornton, J. M. (1993) Procheck—A program to check the stereochemical quality of protein structures. *J. Appl. Crystallogr.* 26, 283–291.
- Vaguine, A. A., Richelle, J., and Wodak, S. J. (1999) SFCHECK: A unified set of procedures for evaluating the quality of macromolecular structure-factor data and their agreement with the atomic model. *Acta Crystallogr., Sect. D* 55, 191–205.
- Lovell, S. C., Davis, I. W., Adrendall, W. B., de Bakker, P. I. W., Word, J. M., Prisant, M. G., Richardson, J. S., and Richardson, D. C. (2003) Structure validation by C alpha geometry: phi,psi and C beta deviation. *Proteins* 50, 437–450.
- DeLano, W. L. (2002) The PyMOL Molecular Graphics System ([www.pymol.org](http://www.pymol.org)).
- Schlichting, I., Berendzen, J., Chu, K., Stock, A. M., Maves, S. A., Benson, D. E., Sweet, R. M., Ringe, D., Petsko, G. A., and Sligar, S. G. (2000) The catalytic pathway of cytochrome p450cam at atomic resolution. *Science* 287, 1615–1622.
- Nagano, S., and Poulos, T. L. (2005) Crystallographic study on the dioxygen complex of wild-type and mutant cytochrome P450cam. Implications for the dioxygen activation mechanism. *J. Biol. Chem.* 280, 31659–31663.
- Nagano, S., Tosha, T., Ishimori, K., Morishima, I., and Poulos, T. L. (2004) Crystal structure of the cytochrome p450cam mutant that exhibits the same spectral perturbations induced by putidaredoxin binding. *J. Biol. Chem.* 279, 42844–42849.
- Sakurai, K., Shimada, H., Hayashi, T., and Tsukihara, T. (2009) Substrate binding induces structural changes in cytochrome P450cam. *Acta Crystallogr., Sect. F: Struct. Biol. Cryst. Commun.* 65, 80–83.
- Lampe, J. N., Brandman, R., Sivaramakrishnan, S., and Ortiz de Montellano, P. R. (2010) 2D NMR and all-atom molecular dynamics of cytochrome P450 CYP119 reveal hidden conformational substrates. *J. Biol. Chem.* (in press).
- Lampe, J. N., Floor, S. N., Gross, J. D., Nishida, C. R., Jiang, Y., Trnka, M. J., and Ortiz de Montellano, P. R. (2008) Ligand-induced conformational heterogeneity of cytochrome P450 CYP119 identified by 2D NMR spectroscopy with the unnatural amino acid (13)C-p-methoxyphenylalanine. *J. Am. Chem. Soc.* 130, 16168–16169.
- Di Primo, C., Hui Bon Hoa, G., Douzou, P., and Sligar, S. (1990) Mutagenesis of a single hydrogen bond in cytochrome P-450 alters cation binding and heme solvation. *J. Biol. Chem.* 265, 5361–5363.
- Manna, S. K., and Mazumdar, S. (2008) Reversible inactivation of cytochrome P450 by alkaline earth metal ions: Auxiliary metal ion induced conformation change and formation of inactive P420 species in CYP101. *J. Inorg. Biochem.* 102, 1312–1321.
- OuYang, B., Pochapsky, S. S., Pagani, G. M., and Pochapsky, T. C. (2006) Specific effects of potassium ion binding on wild-type and L358P cytochrome P450cam. *Biochemistry* 45, 14379–14388.
- Tosha, T., Yoshioka, S., Takahashi, S., Ishimori, K., Shimada, H., and Morishima, I. (2003) NMR study on the structural changes of cytochrome P450cam upon the complex formation with putidaredoxin. Functional significance of the putidaredoxin-induced structural changes. *J. Biol. Chem.* 278, 39809–39821.

45. Tosha, T., Yoshioka, S., Ishimori, K., and Morishima, I. (2004) L358P mutation on cytochrome P450cam simulates structural changes upon putidaredoxin binding: The structural changes trigger electron transfer to oxy-P450cam from electron donors. *J. Biol. Chem.* *279*, 42836–42843.
46. Lipscomb, J. D. (1980) Electron paramagnetic resonance detectable states of cytochrome P-450cam. *Biochemistry* *19*, 3590–3599.
47. Unno, M., Christian, J. F., Benson, D. E., Gerber, N. C., Sligar, S. G., and Champion, P. M. (1997) Resonance Raman investigations of cytochrome P450(cam) complexed with putidaredoxin. *J. Am. Chem. Soc.* *119*, 6614–6620.
48. Aoki, M., Ishimori, K., Fukada, H., Takahashi, K., and Morishima, I. (1998) Isothermal titration calorimetric studies on the associations of putidaredoxin to NADH-putidaredoxin reductase and P450cam. *Biochim. Biophys. Acta* *1384*, 180–188.
49. Furukawa, Y., and Morishima, I. (2001) The role of water molecules in the association of cytochrome P450cam with putidaredoxin. An osmotic pressure study. *J. Biol. Chem.* *276*, 12983–12990.
50. Pochapsky, S. S., Pochapsky, T. C., and Wei, J. W. (2003) A model for effector activity in a highly specific biological electron transfer complex: The cytochrome P450(cam)-putidaredoxin couple. *Biochemistry* *42*, 5649–5656.
51. OuYang, B., Pochapsky, S. S., Dang, M., and Pochapsky, T. C. (2008) A functional proline switch in cytochrome P450cam. *Structure* *16*, 916–923.
52. Davydov, D. R., Hui Bon Hoa, G., and Peterson, J. A. (1999) Dynamics of protein-bound water in the heme domain of P450BM3 studied by high-pressure spectroscopy: comparison with P450cam and P450 2B4. *Biochemistry* *38*, 751–761.
53. Tschirret-Guth, R. A., Koo, L. S., Hoa, G. H., and Ortiz de Montellano, P. R. (2001) Reversible pressure deformation of a thermophilic cytochrome P450 enzyme (CYP119) and its active-site mutants. *J. Am. Chem. Soc.* *123*, 3412–3417.
54. Hui Bon Hoa, G., Di Primo, C., Geze, M., Douzou, P., Kornblatt, J. A., and Sligar, S. G. (1990) The formation of cytochrome P-450 from cytochrome P-420 is promoted by spermine. *Biochemistry* *29*, 6810–6815.
55. Martinis, S. A., Blanke, S. R., Hager, L. P., Sligar, S. G., Hoa, G. H., Rux, J. J., and Dawson, J. H. (1996) Probing the heme iron coordination structure of pressure-induced cytochrome P420cam. *Biochemistry* *35*, 14530–14536.
56. Hanson, L. K., Eaton, W. A., Sligar, S. G., Gunsalus, I. C., Gouterman, M., and Connell, C. R. (1976) Origin of anomalous Soret spectra of carboxycytochrome P-450. *J. Am. Chem. Soc.* *98*, 2672–2674.
57. Jung, C. (1985) Quantum chemical explanation of the hyper spectrum of the carbon-monoxide complex of cytochrome-P-450. *Chem. Phys. Lett.* *113*, 589–596.
58. Perera, R., Sono, M., Sigman, J. A., Pfister, T. D., Lu, Y., and Dawson, J. H. (2003) Neutral thiol as a proximal ligand to ferrous heme iron: Implications for heme proteins that lose cysteine thiolate ligation on reduction. *Proc. Natl. Acad. Sci. U.S.A.* *100*, 3641–3646.
59. Yao, H., McCullough, C. R., Costache, A. D., Pallela, P. K., and Sem, D. S. (2007) Structural evidence for a functionally relevant second camphor binding site in P450cam: Model for substrate entry into a P450 active site. *Proteins* *69*, 125–138.